trans-2-Aryl-*N*,*N*-dipropylcyclopropylamines: Synthesis and Interactions with 5-HT_{1A} Receptors

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Twelve *N*,*N*-dipropyl-substituted derivatives of *trans*-2-arylcyclopropylamine have been prepared and assayed for their ability to displace [³H]-8-OH-DPAT from rat brain 5-HT_{1A} receptors. The new derivatives include phenyl (**7a**), bromo- (**7b**) and fluorophenyl (**7c**-**e**), 2-methoxy-5fluorophenyl (**7h**), and 2-hydroxy-5-fluorophenyl (**7l**) as well as trifluoromethylphenyl (**7f**) and 2,3-dichlorophenyl (**7g**) analogues. In the present series of compounds, electron-withdrawing substituents in the phenyl ring appear to decrease the affinity for 5-HT_{1A} receptors. In contrast, electron-rich aryl groups, such as 2- or 3-thienyl (**7j** and **7k**, respectively), provide compounds with high affinity. The additional bulk produced by the aromatic moiety in the 2-benzothienyl derivative **7i** appears to be detrimental to 5-HT_{1A} receptor affinity. The racemic mixtures of the interesting **7j** and **7l** were resolved into the enantiomers; **7j** and **7l** exhibited a high enantiomeric 5-HT_{1A} receptor affinity ratio (75-fold and 100-fold, respectively). The enantiomers of **7j** and **7l** were evaluated *in vivo* by use of biochemical and behavioral tests in rats. Compound (1*R*,2*R*)-**7j** behaved as a partial agonist whereas (1*R*,2*S*)-**7l** appeared as an efficacious 5-HT_{1A} receptor agonist, stimulating both autoreceptors and postsynaptic receptors.

Introduction

The putative involvement of serotonergic 5-HT_{1A} receptors in anxiety and depression has generated a considerable interest in the medicinal chemistry of selective 5-HT_{1A} receptor agonists and antagonists.¹ A large number of 2-aminotetralin and arylpiperazine derivatives display high affinity and selectivity for the 5-HT_{1A} receptors.² The 2-arylcyclopropylamine derivatives constitute another structural class of 5-HT_{1A} ligands that has been much less investigated. However, trans-cyclopropylamines 1 and 2 have been identified as potent and selective 5-HT_{1A} receptor agonists.³ In contrast to the prototypal 5- HT_{1A} receptor agonist 8-hydroxy-2-(dipropylamino)tetralin (8-OH-DPAT, 3),⁴ which exhibits a low stereoselectivity in its interactions with 5-HT_{1A} receptors, phenylcyclopropylamine derivative 1 appears to be highly stereoselective; the affinity and potency of (1R, 2S)-1 is much higher than that of the antipode.^{3,5}



Our previous study of structure–activity relationships (SAR) of *trans*-2-arylcyclopropylamine derivatives³ was

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focused on variations of the phenolic substitution pattern and of the alkyl groups on the amino group. In the present study we have prepared a series of 12 racemic derivatives (**7a**–**I**) in which the *N*,*N*-dipropylcyclopropylamine moiety has been kept constant and in which the aromatic part has been varied. We have also prepared the enantiomers of the 5-fluoro-2-methoxyphenyl (**7h**), 5-fluoro-2-hydroxyphenyl (**7l**), and 2-thienyl (**7j**) derivatives. The compounds were studied for their ability to displace [³H]-8-OH-DPAT from 5-HT_{1A} receptors. In addition, four selected derivatives were evaluated *in vivo* by use of biochemical and behavioral studies in rats.

Chemistry

The synthesis of the test compounds is described in Scheme 1; *trans* arylated methyl acrylates **4** were prepared from the corresponding aryl iodides by a palladium-catalyzed coupling with methyl acrylate under phase transfer conditions.⁶

4-Fluoro-2-iodoanisole is not commercially available. Attempts to prepare the compound regioselectively failed; we never achieved a regiospecific formation of an anion in the 2-position of 4-fluoroanisole and the 2and 3-halogenated regioisomers were nonseparable. The best regioselectivity was accomplished at room temperature, but the yield was unacceptably low at this temperature. Consequently, an alternative synthetic route was chosen:^{7a} 4-Fluoroanisole was chloromethylated in the 2-position with dimethoxymethane in a hydrochloric acid-sulfuric acid mixture, and the resulting intermediate was treated with hexamethylenetetraamine and acetic acid to form 4-fluoro-2-methoxybenzaldehyde which was converted into the corresponding trans-phenylpropenoate (4h) by a Knoevenagel condensation with malonic acid followed by esterification with methanol (Scheme 2).

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Scheme 1^a



^a Reagents: (a) methyl acrylate, Pd(OAc)₂, NBu₄Cl, K₂CO₃, DMF (method I); (b) (i) CH₂N₂, Pd(OAc)₂, (ii) NaOH, H₂O, MeOH (method III); (c) DPPA, NEt₃, *t*-BuOH (method V); (d) (i) NEt₃, EtOCOCl, acetone, (ii) NaN₃, (iii) *t*-BuOH, (iv) HCl (method VI); (e) (i) NEt₃, EtOCOCl, acetone, (ii) NaN₃, (iii) TMS(CH₂)₂OH, (iv) NBu₄F, THF (method VII); (f) PrI, K₂CO₃, acetonitrile (method VIII).





^{*a*} Reagents: (a) (i) *n*-BuLi, THF, $-78 \rightarrow -25$ °C, (ii) I₂, -78 °C → room temperature; (b) (i) concentrated HCl/concentrated H₂SO₄, 70 °C, (ii) hexamethylenetetramine, 50% aqueous HOAc, reflux; (c) (i) malonic acid, piperidine/pyridine, room temperature → 60 °C, (ii) Δ (100 °C), (iii) catalytic H₂SO₄, MeOH, reflux.

The arylated methyl acrylate derivatives were cyclopropanated with diazomethane in the presence of 0.005 equiv of palladium(II) acetate.⁸ Higher concentrations of the catalyst often led to precipitation of palladium-(0) with accompanying termination of the reaction. Alkaline hydrolysis gave the desired trans-arylcyclopropanecarboxylic acids 5 (Scheme 1). These acids were transformed to the corresponding primary amines by Curtius rearrangements; the one-pot procedure with diphenyl phosphorazidate⁹ worked well in many cases but occasionally produced a complex reaction mixture. Therefore, an alternative method was used for most of the compounds that involved preparation of a mixed anhydride which was treated with sodium azide. The resulting acyl azides were transformed into *tert*-butyl carbamates via the the corresponding isocyanates. These carbamates were hydrolyzed with 1 M hydrochloric acid. However, when the aryl moiety was acid sensitive we prepared the 2-(trimethylsilyl)ethyl carbamate, which was hydrolyzed with tetrabutylammonium fluoride.¹⁰ Due to decomposition of the aryl moiety, we were not able to isolate the desired product

Scheme 3^a



^{*a*} Reagents: (a) (i) NaOH, MeOH, H₂O, (ii) C₂O₂Cl₂, (iii) L-(+)-camphorsultam sodium salt; (b) CH₂N₂, Pd(OAc)₂; (c) (i) Ti(*i*-PrO)₄, BzOH, (ii) 2 M NaOH, MeOH, THF (method IV).

from the hydrolysis (tetrabutylammonium fluoride or iodotrimethylsilane¹¹) of the carbamate function in the 2-furylcyclopropyl derivative.

The *trans* stereochemistry of the primary amines (6) was confirmed by analysis of vicinal proton–proton coupling constants of the cyclopropyl moiety ($J_{H1-H2} \sim 3.5-3.8$ Hz). Compounds 6 were N,N-dipropylated by treatment with 1-iodopropane to afford the corresponding tertiary amines (7). Methoxy derivative 7h was O-demethylated by treatment with hydrobromic acid to form the phenol 7l.

In order to obtain (1.S,2.R)-71 and (1.S,2.S)-7j, we prepared the (+)-enoyl sultams from L-(+)-camphorsultam and the methyl acrylate derivatives **4h** and **4j** (see Scheme 3; the D-(-)-camphorsultam was used in the preparation of the antipodes). The enoyl sultams were cyclopropanated, and the products were recrystallized to provide the corresponding cyclopropanoyl derivatives of high diastereomeric purities.⁷ The camphorsultam moiety of the diastereoisomers was removed by a twostep procedure involving (a) treatment with titanium-(IV) isopropoxide in benzyl alcohol and (b) basepromoted hydrolysis of the resulting benzyl ester, thus providing the enantiomers of **5h** and **5j**. Test compounds were prepared from the enantiomers of **5h** and **5j** according to methods described above.

Physical data of the new compounds are presented in Tables 1 and 2.

Pharmacology

Affinity for 5-HT_{1A} Receptors in Vitro. The affinities of the *N*,*N*-dipropylcyclopropylamine derivatives for rat hippocampal and cortical 5-HT_{1A} receptors were determined in competition studies with [³H]-8-OH-DPAT. In addition, the structurally related and previously reported³ enantiomers of **1**, **2**, and **10** were evaluated in the same assay for comparison. The K_i values are reported in Table 3.

In Vivo Biochemistry. The rationale for the *in vivo* biochemical experimental protocol is based on the wellestablished phenomenon of (auto)receptor-mediated feedback inhibition of presynaptic monoaminergic neurones.^{12–15} Thus, the synthesis rate of the catecholamines dopamine (DA) and norepinephrine (NE) is inhibited by agonists via an action at DA receptors and α -adrenoceptors, respectively. Similarly, the synthesis rate of serotonin (5-HT) is inhibited by 5-HT

Table 1. Physical Data of Some Substituted Methyl Propenoates

			Ar´			
compd	Ar	prepn ^a method	yield, %	mp, °C	bp, °C/mmHg	formula
4b	2-BrPh	Ι	80	oil ^b		C ₁₀ H ₉ BrO ₂
4 c	2-FPh	I	84		122-123/9.5 ^{c,d}	$C_{10}H_9FO_2$
4d	3-FPh	I	73		68-70/0.05 ^{c,d}	$C_{10}H_9FO_2 \cdot 1/_8H_2O$
4e	4-FPh	I	92		114-116/4.5 ^{c,d}	$C_{10}H_9FO_2$
4f	2-CF₃Ph	I	78		66-68/0.1 ^c	$C_{11}H_9F_3O_2$
4g	2,3-(Cl) ₂ Ph	I	88	69 - 71		$C_{10}H_8Cl_2O_2$
4h	5-F,2-OCH₃Ph	II	94	43 - 44		$C_{11}H_{11}FO_3$
4i	benzothienyl	I	75	$124 - 125.5^{e}$		$C_{12}H_{10}O_2S$
4j	2-thienyl	I	96	$46 - 47^{f}$		$C_8H_8O_2S$
4 k	3-thienyl	Ι	50		70-72/0.08 ^{c,g}	$C_8H_8O_2S\boldsymbol{\cdot}^{1\!/}_8H_2O$

COOCH3

^a See Experimental Section. ^b Reference 30. ^c Purified by distillation. ^d Reference 31. ^e Reference 32. ^f Reference 33. ^g Reference 34.

receptor agonists. The accumulation of 3,4-dihydroxyphenylalanine (DOPA), following decarboxylase inhibition by means of 3-hydroxybenzylhydrazine (NSD1015; 100 mg/kg sc), was used as an index of the DA synthesis rate in the DA-rich parts (limbic forebrain, striatum) and of the NE synthesis rate in the remaining NEpredominated portions (mainly cortex). The accumulation of 5-hydroxytryptophan (5-HTP) following decarboxylase inhibition was taken as an indicator of the 5-HT synthesis rate in all three brain parts (Figure 1). Reserpine-pretreated rats (5 mg/kg, 18–20 h before) were used throughout this study. Reserpine depletes central monoamine stores, thereby facilitating the assessment of direct receptor agonist properties of the test compounds, both with regard to biochemical and behavioral effects.¹⁶

The (1*R*,2*R*)-enantiomer of 7j and the (1*R*,2*S*)-enantiomer of 7l decreased the cerebral 5-HT synthesis rate by about 50% from control values at the highest doses tested (35 μ mol/kg sc) (Figure 1). This is similar to the maximum 5-HT synthesis reduction elicited by (1R, 2S)- $\mathbf{1}^{3}$ and the reference 5-HT_{1A} receptor agonist $\mathbf{3}^{4}$ A rough graphical approximation of the doses required to produce a half-maximal suppression of the 5-HT synthesis indicates that (1R,2S)-71-the 5-fluorinated derivative of (1R, 2S)-**1**—and the thienyl analogue (1R, 2R)-**7j** display similar potency (\sim 1.5–3 µmol/kg), both being about 5–10 times less potent than (1*R*,2*S*)-1 (cf. ref 3). None of the compounds tested caused any significant change in rat brain DOPA accumulation (data not shown). This was true in both DA-rich (limbic forebrain, striatum) and NA-predominated (cortex) brain areas, and suggests lack of potent DA and NA a-receptor stimulatory properties, respectively.

5-HT Syndrome and Gross Behavioral Observations. Stimulation of postsynaptic 5-HT_{1A} receptors by means of, for example, administration of 3 results in a clear-cut motor behavioral syndrome, the "5-HT syndrome", in rats. Flattened body posture, forepaw treading, and hindlimb abduction are the most prominent features of this syndrome.^{4c,17} The ability of the test drugs to elicit putative postsynaptic 5-HT_{1A} receptor agonist properties was thus assessed by rating the occurrence and intensity of these behavioral components in the reserpinized rats (Table 4).¹⁶ The gross behavior of the animals was observed in their home cages. Immediately after the scoring session the rats were injected with NSD1015 and subsequently used for determination of brain 5-HT and catecholamine synthesis rates (cf. above). (1S, 2S)-7j and (1R, 2S)-1 both induced a clear-cut 5-HT behavioral syndrome, qualitatively similar to that seen after the reference 5-HT_{1A} receptor agonist **3**. In contrast, the effects of (1R,2R)-**7j** were at best mild, even at the highest dose tested $(35 \,\mu$ mol/kg sc; Table 4). (1.5,2R)-**7l** failed to induce any appreciable behavioral change under these conditions.

Discussion

The previously prepared (1R,2S)-enantiomers of 2and 3-hydroxy derivatives **1** and **2** had 177- and 29-fold higher 5-HT_{1A} receptor affinity, respectively, than the (1S,2R)-antipodes (Table 3). This is consistent with earlier studies of the *in vivo* 5-HT_{1A} receptor agonist properties of these derivatives.³ The 2-methoxyphenyl derivative (1R,2S)-**10** was only about 3-fold less potent than the phenolic analogue (1R,2S)-**1**. A similar lowaffinity ratio has been reported between the 8-hydroxy and 8-methoxy derivatives of 2-(dipropylamino)tetralin (**3** and **11**, respectively).¹⁸ Additional comparisons with



the 2-aminotetralin series can be made: In studies of C8-substituted 2-(dipropylamino)tetralin derivatives it has been demonstrated that most of the affinity for 5-HT_{1A} receptors is retained when the C8-substituent is replaced by a hydrogen.¹⁸ A similar trend was observed in the present study since phenyl derivative **7a** is about 5-fold less potent than the 2-hydroxyphenyl analogue (1*R*,2*S*)-**1** (Table 3). These findings point to similarities in the SAR of the two series and indicates that the modes of binding of the 2-aminotetralin and arylcyclopropylamine derivatives to the 5-HT_{1A} receptors are similar (compare the SAR discussion in refs 19 and 20).

When the affinities of the halogen-substituted derivatives are compared with that of the phenyl derivative **7a**, it is apparent that halogen substitution in the phenyl ring does not affect (2-Br, **7b**), decreases (2-F, 3-F, or 4-F; **7c**, **7d**, or **7e**, respectively), or abolishes (2,3- Cl_2 ; **7g**) the affinity for 5-HT_{1A} receptors (Table 3). Introduction of the strongly electron withdrawing trifluoromethyl group in the C2-position of the phenyl moiety, affording **7f**, also abolished affinity. These data indicate that the introduction of electron-withdrawing substituents in the phenyl ring is detrimental for affinity. In contrast, hydroxyl groups, which enhance

				Ar			
compd	Ar	Y	prepn meth	yield, %	mp, °C	recrystn solvents ^a	formula
5a	Ph	COOH	III^{b}	91	89.5-90.5 ^c	\mathbf{NR}^d	$C_{10}H_{10}O_2$
5b	2-BrPh	COOH	III	75	106 - 107.5	NR	$C_{10}H_9BrO_2$
5c	2-FPh	COOH	III	92	oil		$C_{10}H_9FO_2 \cdot \frac{1}{8}H_2O$
5d	3-FPh	COOH	III	90	80-81	NR	$C_{10}H_9FO_2$
5e	4-FPh	COOH	III	94	110-112	NR	$C_{10}H_9FO_2$
5f	2-CF ₃ Ph	COOH	III	69	117 - 118.5	F	$C_{11}H_9F_3O_2$
5g	2,3-(Cl) ₂ Ph	COOH	III	63	129.5 - 130.5	NR	$C_{10}H_8Cl_2O_2$
5h	5-F,2-OCH ₃ Ph	COOH	III	89	165.5 - 166.5	NR	$C_{11}H_{11}FO_3$
(1 <i>S</i> ,2 <i>S</i>)- 5h	5-F,2-OCH ₃ Ph ^e	COOH	IV	93	125.5 - 126	NR	$C_{11}H_{11}FO_3$
(1 <i>R</i> ,2 <i>R</i>)- 5h	5-F,2-OCH ₃ Ph ^f	COOH	IV	92	125.5 - 126	NR	$C_{11}H_{11}FO_3$
5i	benzothienyl	COOH	III	84	$165.5 - 167^{g}$	С	$C_{12}H_{10}O_2S$
5j	2-thienyl	COOH	III	87	$46 - 48^{h}$	NR	$C_8H_8O_2S$
(1 <i>S</i> ,2 <i>S</i>)- 5j	2-thienyl ⁱ	СООН	IV	60	oil		$C_8H_8O_2S$
(1 <i>R</i> ,2 <i>R</i>)- 5j	2-thienyl⁄	СООН	IV	78	oil		$C_8H_8O_2S$
5k	3-thienyl	COOH	III	91	73 - 75	NR	$C_8H_8O_2S$
6a	Ph	NH_2	V	45	154–156 ^c	Α	C ₉ H ₁₁ N·HCl
6b	2-BrPh	NH_2	VI	45	172 - 174	Α	C ₉ H ₁₀ BrN·HCl
6c	2-FPh	NH_2	V	40	$130 - 133^{k}$	Α	C ₉ H ₁₀ FN·HCl
6d	3-FPh	NH_2	VI	85	$181 - 184^{k}$	Α	C ₉ H ₁₀ FN·HCl
6e	4-FPh	NH_2	VI	68	190–192 ^{<i>k</i>,1}	m	C ₉ H ₁₀ FN·HCl
6f	2-CF₃Ph	NH_2	VI	63	177 - 179	А	$C_{10}H_{10}F_3N\cdot HCl$
6g	2,3-(Cl) ₂ Ph	NH_2	V	43	193 - 195	А	C ₉ H ₉ Cl ₂ N·HCl· ¹ / ₄ H ₂ O
6h	5-F,2-OCH₃Ph	NH_2	VI	58	201.5 - 203.5	В	C ₁₀ H ₁₂ FNO·HCl
(1 <i>S</i> ,2 <i>R</i>)- 6h	5-F,2-OCH ₃ Ph ⁿ	NH_2	VI	58	213.5 - 216	А	$C_{10}H_{12}FNO\cdot HCl$
(1 <i>R</i> ,2 <i>S</i>)- 6h	5-F,2-OCH ₃ Ph ^o	NH_2	VI	59	215 - 218	А	$C_{10}H_{12}FNO\cdot HCl$
6i	benzothienyl	NH_2	VI	82	190 ^{g,1}	А	$C_{11}H_{11}NS \cdot C_2H_2O_4 \cdot \frac{1}{4}H_2O$
6j	2-thienyl	NH_2	VII	36	$174 - 175^{g,I}$	А	$C_7H_9NS \cdot \frac{1}{2}C_2O_4H_2$
(1 <i>S</i> ,2 <i>S</i>)- 6j	2-thienyl ^p	NH_2	VII	58	178-180	А	$C_7H_9NS \cdot 1/_2C_2O_4H_2$
(1 <i>R</i> ,2 <i>R</i>)- 6j	2-thienyl ^q	NH_2	VII	49	$207 - 210^{1}$	А	$C_7H_9NS \cdot C_2O_4H_2$
6k	3-thienyl	NH_2	VII	35	$171 - 172^{1}$	A	$C_7H_9NS \cdot C_2O_4H_2$
7a	Ph	$N(n-C_{3}H_{7})_{2}$	VIII	56	154 - 155	A	$C_{15}H_{23}N\cdot HCl$
7b	2-BrPh	$N(n-C_{3}H_{7})_{2}$	VIII	75	148.5 - 150.5	А	$C_{15}H_{22}BrN \cdot HCl$
7c	2-FPh	$N(n-C_{3}H_{7})_{2}$	VIII	41	93-94	D	$C_{15}H_{22}FN\cdot HCl$
7d	3-FPh	$N(n-C_{3}H_{7})_{2}$	VIII	52	148 - 151	E	$C_{16}H_{23}N\cdot HCl$
7e	4-FPh	$N(n-C_{3}H_{7})_{2}$	VIII	63	117 - 119	D	$C_{16}H_{23}N\cdot HCl$
7f	2-CF ₃ Ph	N(<i>n</i> -C ₃ H ₇) ₂	VIII	33	130 - 132	A	$C_{16}H_{22}F_{3}N \cdot HCl$
7g	2,3-(Cl) ₂ Ph	N(<i>n</i> -C ₃ H ₇) ₂	VIII	74	141 - 143	A	$C_{15}H_{21}Cl_2N\cdot HCl$
7h	5-F,2-OCH ₃ Ph	N(<i>n</i> -C ₃ H ₇) ₂	VIII	75	154.5 - 156	A	C ₁₆ H ₂₄ FNO·HCl
(1 <i>S</i> ,2 <i>R</i>)- 7h	5-F,2-OCH ₃ Ph ^r	$N(n-C_3H_7)_2$	VIII	42	163 - 165	A	$C_{16}H_{24}FNO\cdot HCI$
(1 <i>R</i> ,2 <i>S</i>)-7h	5-F,2-OCH ₃ Ph ^s	$N(n-C_3H_7)_2$	VIII	48	163.5 - 165	A	$C_{16}H_{24}FNO\cdot HCI$
7i	benzothienyl	$N(n-C_3H_7)_2$	VIII	41	132 - 133.5	A	$C_{17}H_{23}NS \cdot HCI \cdot \frac{1}{4}H_2O$
7j	2-thienyl	$N(n-C_3H_7)_2$	VIII	94	83-85	A	$C_{13}H_{21}NS \cdot C_2O_4H_2 \cdot I_4H_2O_4$
(1 <i>S</i> ,2 <i>S</i>)- 7 j	2-thienyl ^{<i>t</i>}	$N(n-C_3H_7)_2$	VIII	56	$172 - 175^{\prime}$	A	$C_{13}H_{21}NS \cdot HCI \cdot \frac{1}{8}H_2O$
(1 <i>R</i> ,2 <i>R</i>)- 7j	2-thienyl ^{<i>u</i>}	$N(n-C_3H_7)_2$	VIII	80	$170 - 172^{1}$	A	$C_{13}H_{21}NS \cdot HCI \cdot \frac{1}{4}H_2O$
7k	3-thienyl	$N(n-C_3H_7)_2$	VIII	68	150-52	A	$C_{13}H_{21}NS \cdot HCI$
71	5-F,2-OHPh	$N(n-C_3H_7)_2$	IX	65	167.5-169	A	C ₁₅ H ₂₂ FNO•HCl
(1 <i>S</i> ,2 <i>R</i>)- 71 (1 <i>R</i> ,2 <i>S</i>)- 71	5-F,2-OHPh ^v 5-F,2-OHPh ^w	N(<i>n</i> -C ₃ H ₇) ₂ N(<i>n</i> -C ₃ H ₇) ₂	IX IX	52 58	$169 - 172 \\ 170 - 174$	A A	C ₁₅ H ₂₂ FNO·HCl C ₁₅ H ₂₂ FNO·HCl

^{*a*} A = methanol/ether, B = ethanol/ether, C = toluene, D = ether (after several days at -25 °C), E = 2-propanol/ether, F = ether. ^{*b*} Prepared from the commercially available methyl cinnamate. ^{*c*} Reference 35. ^{*d*} Not recrystallized due to high initial purity. ^{*e*} [α]²²_D = +211 (c = 1.0, CH₂Cl₂). ^{*f*} [α]²²_D = -212 (c = 1.0, CH₂Cl₂). ^{*g*} Reference 36. ^{*h*} Reference 37. ^{*i*} [α]²²_D = +328 (c = 1.0, CH₂Cl₂). ^{*j*} [α]²²_D = -326 (c = 1.1, CH₂Cl₂). ^{*k*} Reference 38. ^{*l*} Decomposed. ^{*m*} Not recrystallized, decomposed on attempt to recrystallize from methanol/ether. ^{*n*} [α]²²_D = +42.0 (c = 1.0, MeOH). ^{*o*} [α]²²_D = -41.3 (c = 1.0, MeOH). ^{*p*} [α]²²_D = +77.4 (c = 0.6, MeOH). ^{*q*} [α]²²_D = -57.0 (c = 0.9, MeOH), [α]²²_D = -58.8 (c = 0.5, MeOH). ^{*r*} [α]²²_D = +8.9 (c = 1.0, MeOH). ^{*s*} [α]²²_D = -9.4 (c = 1.0, MeOH). ^{*t*} [α]²²_D = +75.2 (c = 1.0, MeOH). ^{*u*} [α]²²_D = -76.3 (c = 1.1, MeOH). ^{*v*} [α]²²_D = -7.8 (c = 1.0, MeOH).

the electron density of the aromatic ring, appear to be beneficial for 5-HT_{1A} receptor affinity (cf. above).

Three compounds with electron-rich heteroaromatic rings, 2-thienyl (**7j**), 3-thienyl (**7k**), and 2-benzothienyl (**7i**), were synthesized and tested as potential bioisosteres of the potent phenolic derivatives **1** and **2** (Table 3). The 2-thienyl derivative **7j** proved to be a potent 5-HT_{1A} receptor ligand, the affinity being 2–3-fold less than that of the phenolic analogues **1** and **2**. The affinity of the 3-thienyl analogue **7k** was slightly lower. It thus appears that the 2- and 3-thienyl rings constitute good isosteric replacements for the 2- and 3-hydroxyphenyl moieties in **1** and **2**. The 2-benzothienyl derivative **7i** was inactive, most likely due to the steric bulk of the fused phenyl moiety.

Because of the high affinity of racemic **7j**, we synthesized and tested the enantiomers. As might have been predicted on the basis of previously observed stereoselectivities in the phenylcyclopropylamine series, ^{3,5} **7j** exhibited a large enantiomeric affinity ratio, the affinity of the (1*R*,2*R*)-enantiomer being 75-fold higher than that of the antipode. In the *ex vivo* biochemical experiments, (1*R*,2*R*)-**7j** behaved as a full 5-HT_{1A} receptor agonist but of slightly lower potency than (1*R*,2*S*)-**1**. In contrast, the antipode, (1*S*,2*S*)-**7j**, was unable to reduce the 5-HTP levels even at the highest dose level (35 μ mol/ kg) (Figure 1). It is noteworthy that neither enantiomer of **7j** induced a clear-cut 5-HT behavioral syndrome (Table 4) in the reserpinized rats. Thus, the high 5-HT_{1A} receptor affinity of **7j** (Table 2) was accompanied

Table 3. Affinities for 5-HT_{1A} Receptors Defined by the Displacement of [³H]-8-OH-DPAT

compd	<i>K</i> _i , nM	range	n _H
7a	24	(21-28)	0.88
7b	22	(20 - 23)	0.87
7c	59	(48-77)	0.94
7d	100	(85 - 130)	0.74
7e	76	(63-95)	0.83
7f	>1000		
7g	>300		
7h	140	(89-280)	0.52
(1 <i>S</i> ,2 <i>R</i>)- 7h	199	(174 - 224)	0.90
(1 <i>R</i> ,2 <i>S</i>)- 7h	60	(39-87)	0.80
7i	>1000		
7j	13	(12 - 14)	0.94
(1 <i>S</i> ,2 <i>S</i>)-7j	820	(710-980)	1.00
(1 <i>R</i> ,2 <i>R</i>)- 7 j	11	(10-13)	0.96
7k	21	(19–23)	0.93
71	99	(78 - 140)	0.84
(1 <i>S</i> ,2 <i>R</i>)- 71	348	(184 - 503)	0.90
(1 <i>R</i> ,2 <i>S</i>)- 71	3.2	(2.7 - 3.6)	0.88
(1 <i>S</i> ,2 <i>R</i>)- 1	140	(100 - 230)	0.90
(1 <i>R</i> ,2 <i>S</i>)- 1	4.9	(4.6 - 5.2)	0.97
(1 <i>S</i> ,2 <i>R</i>)- 2	460	(240-760)	0.70
(1 <i>R</i> ,2 <i>S</i>)- 2	2.6	(2.2 - 3.2)	0.80
(1 <i>S</i> ,2 <i>R</i>)- 10	230	(160-430)	0.90
(1 <i>R</i> ,2 <i>S</i>)- 10	17	(16-20)	0.75

by potent agonist properties in the *ex vivo* biochemical model, but low apparent efficacy (if any) in the behavioral test model, reflecting, respectively, the ability to activate synthesis controlling 5-HT_{1A} autoreceptors and postsynaptic 5-HT_{1A} receptors mediating the 5-HT behavioral syndrome in reserpinized rats. These data are consistent with a characterization of (1R,2R)-**7j** as a partial 5-HT_{1A} receptor agonist. Thus, the isosteric replacement of the hydroxyphenyl moiety of (1S,2R)-**1** with a 2-thienyl ring appears to result in a considerable loss in efficacy, manifested under the present conditions only at the level of postsynaptic 5-HT_{1A} receptors, likely due to differences in apparent receptor reserve (cf. ref 21). In contrast, most of the affinity is retained.



Introduction of a C5-fluoro substituent in 2-aminotetralin derivative 3, producing 12 (UH-301),²² reduces mainly efficacy but also affinity. Thus, (R)-12 is a partial 5-HT_{1A} receptor agonist whereas (R)-3 is a full agonist, and (S)-12 is a 5-HT_{1A} receptor antagonist whereas (S)-3 is a partial agonist.⁵ Similar effects were not observed when a C5-fluoro substituent was introduced in the cyclopropylamine-derived agonist (1R, 2S)-1; (1R,2S)-71 had about the same affinity as (1R,2S)-1 and displayed clear-cut in vivo biochemical (Figure 1) as well as behavioral (Table 4) agonist activity. (1S,2R)-71, on the other hand, was 100-fold less potent than the antipode in terms of affinity and only mildly affected 5-HTP accumulation at the highest dose tested (35 μ mol/ kg sc). Similarly, (1S,2R)-71 failed to produce a fullblown 5-HT behavioral syndrome.

Experimental Section

Chemistry. General Comments. Melting points (uncorrected) were determined in open glass capillaries on a Thomas-Hoover apparatus. ¹H NMR spectra were recorded on a JEOL JNM-EX270 spectrometer at 270 MHz, and ¹³C NMR spectra were recorded on a JEOL FX 90Q spectrometer at 22.5 MHz, except where noted. NMR spectra were referenced to internal tetramethylsilane. GLC analysis was performed on a Shimadzu GČ14A equipped with a FID detector and a HP-1 column (50 m \times 0.32 mm). Mass spectral analysis was performed on a Hewlett-Packard mass spectrometer HP 5971A MSD connected to a gas chromatograph HP GC 5890 Series 2, equipped with a HP-1 column (25 m \times 0.2 mm). Optical rotations were obtained on a Perkin-Elmer 241 polarimeter. The elemental analyses (C, H, and N), which were performed by Micro Kemi AB, Uppsala, Sweden, were within 0.4% of the theoretical values. TLC analysis was carried out on aluminum sheets precoated with silica gel F_{254} (0.2 mm) or on Al_2O_3 60 F₂₅₄ neutral (Typ E), E. Merck, and was visualized with UV light and/or I₂. All reactions except the cyclopropanations were performed under nitrogen.

2-Iodobenzothiophene.²³ Benzothiophene (10 g, 75 mmol) was dissolved in ether (100 mL). The solution was cooled to 0 °C, and butyllithium (1.6 M in hexane, 49 mL, 78 mmol) was added. After 3.5 h the solution was cooled to -50 °C, iodine (18.9 g, 75 mmol) was added, and the reaction temperature was slowly increased to room temperature. Saturated aqueous NH₄Cl (50 mL) and saturated aqueous Na₂S₂O₃ (70 mL) were added to the mixture. The organic layer was separated, washed with H_2O (2 × 100 mL), dried (MgSO₄), filtered, and concentrated. The crude product was purified by distillation (bp 153-156 °C/14 mmHg). The resulting oil was dissolved in hexane. The solution was treated with charcoal, filtered, and concentrated to afford 8.31 g (43%) of pure 2-iodobenzothiophene as an oil: ¹H NMR (acetone- d_6 , 90 MHz) δ 7.96-7.77 (m, 2 H), 7.71 (s, 1 H), 7.44-7.23 (m, 2 H); ¹³C NMR $(acetone-d_6) \delta 145.0, 141.8, 134.9, 125.4, 125.3, 123.2, 122.1,$ 79.2.

Below are examples of the methods described in Scheme 1: Methyl trans-3-(2-Bromophenyl)propenoate (4b). Method I. A mixture of 2-bromoiodobenzene (7.0 g, 24.7 mmol), methyl acrylate (2.8 mL, 31.9 mmol), tetrabutylammonium chloride (6.9 g, 24.7 mmol), Pd(OAc)₂ (0.11 g, 0.5 mmol), and finely grounded K₂CO₃ (8.55 g, 62 mmol) in DMF (26 mL) was stirred at 50 °C for 19 h. Petroleum ether (150 mL) and brine (40 mL) were added, and the mixture was filtered under suction. The filtrate was collected, and the aqueous layer was extracted with petroleum ether (3 \times 100 mL). The combined organic layers were dried (MgSO₄), filtered, and concentrated. The residue was purified on a silica column (ether/petroleum ether, 1:4) to afford 4.8 g of pure 4b: ¹H NMR (CDCl₃, 90 MHz) δ 8.06 (d, J = 16.0 Hz, 1 H), 7.66– 7.55 (m, 2 H), 7.41–7.10 (m, 2 H), 6.38 (d, J = 16.0 Hz, 1 H), 3.82 (s, 1 H); ¹³C NMR (CDCl₃) δ 166.7, 143.1, 134.4, 133.4, 131.2, 127.7 (2 C's), 125.3, 120.6, 51.8.

Methyl trans-3-(5-Fluoro-2-methoxyphenyl)propenoate (4h). Method II. Concentrated H_2SO_4 (3 mL) was added to a solution of *trans-*3-(5-fluoro-2-methoxyphenyl)propenoic acid^{7a} (6.0 g, 30.6 mmol) in MeOH (150 mL). The solution was heated to reflux for 5 h. Saturated aqueous Na₂CO₃ was added to the resulting solution until neutral pH. The solution was filtered and concentrated. The residue was partitioned between ether (100 mL) and H₂O (150 mL). The organic layer was washed with H₂O (100 mL), dried (MgSO₄), filtered, and concentrated. The residue was purified on a silica column eluted with ether/petroleum ether (1:4) to afford **4h** as a white solid.

trans-2-(2-Fluorophenyl)cyclopropanecarboxylic Acid (5c). Method III. Diazomethane (*CAUTION*)²⁴ was prepared as previously described;²⁴ a solution of *N*-methyl-*N*-nitroso-4toluenesulfonamide (Diazogen, 5.95 g, 27.8 mmol) in ether (60 mL) was slowly added to a heated (70 °C oil bath) mixture of KOH (4.67 g, 83.3 mmol), ether (10 mL), H₂O (50 mL), and 2-(2-ethoxyethoxy)ethanol (50 mL). The resulting ether solution of diazomethane was continuously distilled into a stirred, cooled (-10 °C) solution of **4c** (0.50 g, 2.8 mmol) and Pd(OAc)₂ (3.1 mg, 0.0014 mmol) in CH₂Cl₂/ether (1:2; 75 mL). The reaction was quenched by addition of a few milliliters of HOAc after 1.5 h. The resulting mixture was washed with saturated aqueous NaHCO₃ (2 × 3 mL), dried (MgSO₄), filtered, and





Figure 1. Effect of (1R,2R)-**7j**, (1S,2S)-**7j**, (1R,2S)-**7l**, (1S,2R)-**7l**, and (1R,2S)-**1** on cerebral 5-HT synthesis in rats. Reserpinpretreated rats (5 mg/kg sc, 18–20 h before the experiment) were given 0.35, 3.5, or 35 μ mol/kg sc of (1R,2R)-**7j**, (1S,2S)-**7j**, (1R,2S)-**7l**, (1S,2R)-**7l**, or (1R,2S)-**7l**, or (1R,2S)-**7l**, or (1R,2S)-**7l**, or (1R,2S)-**7l**, or (1R,2S)-**7l**, or (1R,2S)-**1** 60 min, and NSD1015 (100 mg/kg sc) 30 min before death; controls received NaCl and NSD1015 at corresponding time intervals. Shown are the 5-HTP contents (ng/g tissue) in limbic forebrain (a), striatal (b), and cortical (c) brain tissue samples, means ± SEM of two to seven observations. Vertical figures inside the bars depict doses of the test compounds (0.35–35 μ mol/kg sc). * $p \le 0.05$ vs corresponding NaCl control values.

Table 4. 5-HT Behavioral Effects of (1*R*,2*R*)-**7j**, (1*S*,2*S*)-**7j**, (1*R*,2*S*)-**7l**, (1*S*,2*R*)-**7l**, and (1*R*,2*S*)-**1** in Reserpinized Rats^a

	dose, μ mol/kg sc					
0	0.35	3.5	35			
0 (0-0) 7						
	${0 \atop 2}(0{-}0)$	0 (0-0) 3	$\frac{1}{3}(0-1)$			
	~	0 (0-0)				
		0 (0-0)	4 (3-5)			
		1 (1-2)				
		$\frac{5}{4}(2-5)$	3			
	0 0 (0-0) 7		$\begin{tabular}{ c c c c c } \hline & dose, μmol/kg sc \\\hline \hline 0 & 0.35 & 3.5 \\\hline 0 & (0-0) & & \\ 7 & & & \\ 0 & (0-0) & & 0 & (0-0) \\ & & & & 3 \\ 0 & (0-0) & & & \\ & & & & 0 & (0-0) \\ & & & & 3 \\ & & & & 0 & (0-0) \\ & & & & 3 \\ & & & & 1 & (1-2) \\ & & & 5 \\ & & & 4 & (2-5) \\ & & & 3 \\ \hline \end{array}$			

 a Shown are the summed "5-HT behavioral syndrome" scores (median, range in brackets; n observations) obtained in a 60-s rating session, starting 30 min after administration of the test compounds (0.35–35 μ mol/kg sc) in reserpine-pretreated rats (5 mg/kg sc, 18–20 h before experiment).

concentrated. The crude product was purified on a silica column eluted with ether/petroleum ether (1:6). The resulting ester was dissolved in MeOH (10 mL), and 2 M aqueous NaOH (5 mL) was added. The solution was stirred at room temperature for 2 h. The MeOH was evaporated, and the remaining solution was diluted with H_2O (15 mL), washed with ether (20 mL), acidified with 5 M aqueous HCl, and extracted with ether (3 × 20 mL). The combined organic layers were dried (MgSO₄),

filtered, and concentrated to afford pure **5**c: ¹H NMR (CDCl₃) δ 7.24–7.16 (m, 1 H), 7.10–6.95 (m, 3 H), 2.75 (ddd, J_1 = 4.1 Hz, J_2 = 7.0 Hz, J_3 = 9.5 Hz, 1 H), 1.95 (ddd, J_1 = 4.1 Hz, J_2 = 4.8 Hz, J_3 = 8.3 Hz, 1 H), 1.67 (ddd, J_1 = 4.8 Hz, J_2 = 4.7 Hz, J_3 = 9.5 Hz, 1 H), 1.45 (ddd, J_1 = 4.7 Hz, J_2 = 7.0 Hz, J_3 = 8.3 Hz, 1 H), 1.45 (ddd, J_1 = 4.7 Hz, J_2 = 7.0 Hz, J_3 = 8.3 Hz, 1 H), 1.45 (ddd, J_1 = 4.7 Hz, J_2 = 7.0 Hz, J_3 = 8.3 Hz, 1 H); ¹³C NMR (CDCl₃) δ 180.1, 161.7 (d, J = 246.4 Hz), 128.2 (d, J = 8.4 Hz), 127.0 (d, J = 4.2 Hz), 126.1, 124.1 (d, J = 3.5 Hz), 115.4 (d, J = 21.6 Hz), 22.7 (d, J = 1.4 Hz), 20.8 (d, J = 4.9 Hz), 16.2.

(15,25)-trans-2-(2-Thienyl)cyclopropanecarboxylic Acid [(1.S,2.S)-5j]. Method IV. Titanium isopropoxide (6.0 mL, 21 mmol) was added to a solution of (+)-98a (7.41 g, 21 mmol) in benzyl alcohol (40 mL). The solution was heated (150 °C, oil bath) for 45 min and then allowed to cool to room temperature. The solution was concentrated and the residue was purified on a silica column eluted with ether/petroleum ether (1:9), yielding 3.62 g of a mixture of the benzyl and the isopropyl esters. The mixture was dissolved in MeOH (30 mL), and 2 M aqueous NaOH (30 mL) was added. The resulting solution was stirred at room temperature for 6 h and finally heated to 40 °C (oil bath) for 20 min in order to dissolve the remaining substrate. The solution was concentrated, and the residue was washed with CH₂Cl₂ (3 \times 30 mL), EtOAc (3 \times 30 mL), and ether (3 \times 30 mL), acidified with 5 M aqueous HCl, extracted with CH_2Cl_2 (3 \times 30 mL), dried (MgSO₄), filtered, and concentrated to yield 2.11 g (60%) of (1S,2S)-5j: ¹H NMR (CDCl₃) δ 7.11 (dd, $J_1 = 1.3$ Hz, $J_2 = 5.1$ Hz, 1 H), 6.91 (dd, J_1 = 3.6 Hz, J_2 = 5.0 Hz, 1 H), 6.84 (m, 1 H), 2.78 (ddd, J_1 = 4.0 Hz, $J_2 = 6.6$ Hz, $J_3 = 9.2$ Hz, 1 H), 1.94 (ddd, $J_1 = 4.0$ Hz, J_2 = 5.2 Hz, J_3 = 8.4 Hz, 1 H), 1.69 (ddd, J_1 = 4.6 Hz, J_2 = 5.2 Hz, $J_3 = 9.2$ Hz, 1 H), 1.42 (ddd, $J_1 = 4.6$ Hz, $J_2 = 6.6$ Hz, J_3

= 8.4 Hz, 1 H); ¹³C NMR (CDCl₃) δ 179.5, 143.4, 126.9, 124.1, 123.4, 24.8, 22.3, 18.3.

(1*R*,2*R*)-*trans*-2-(2-Thienyl)cyclopropanecarboxylic Acid [(1*R*,2*R*)-5*j*]. The acid (1*R*,2*R*)-5*j* was prepared from (-)- 9^{7a} (6.62 g, 24 mmol) yielding 3.08 g (78%) of pure carboxylic acid.

trans-2-(2,3-Dichlorophenyl)cyclopropylamine (6g). Method V. Diphenyl phosphorazidate (1.93 mL, 9.0 mmol) and triethylamine (1.25 mL, 12 mmol) was added to a solution of 5g (2.00 g, 8.2 mmol) in dry t-BuOH (20 mL). The resulting solution was stirred at 90 °C (bath temperature) for 42 h and was then concentrated. The residue was partitioned between 10% aqueous Na_2CO_3 (40 mL) and ether (4 \times 40 mL). The combined organic layers were dried (Na₂SO₄), filtered, and concentrated. The residue was purified on a silica column eluted with ether/petroleum ether (1:4). The resulting tertbutyl carbamate was treated with 1 M aqueous HCl (50 mL) at 100 °C overnight. The solution was washed with ether (3 \times 40 mL), and saturated aqueous K₂CO₃ was added to pH about 10. The mixture was extracted with EtOAc (3 \times 70 mL). The combined organic layers were dried (K₂CO₃), filtered, and concentrated. The resulting primary amine was converted into the hydrochloride and recrystallized to afford 6g·HCl: ¹H NMR (CD₃OD) δ 7.46 (d, J = 7.9 Hz, 1 H), 7.27 (dd, $J_1 = J_2 = 7.9$ Hz, 1 H), 7.12 (d, J = 7.9 Hz, 1 H), 2.86 (ddd, $J_1 = 3.8$ Hz, J_2 = 4.6 Hz, $J_3 = 7.9$ Hz, 1 H), 2.64 (ddd, $J_1 = 3.8$ Hz, $J_2 = 6.6$ Hz, $J_3 = 10.2$ Hz, 1 H), 1.51 (ddd, $J_1 = 4.6$ Hz, $J_2 = 6.6$ Hz, J_3 = 10.2 Hz, 1 H), 1.41 (ddd, $J_1 = 6.6$ Hz, $J_2 = 6.6$ Hz, $J_3 = 7.9$ Hz, 1 H); ¹³C NMR (CD₃OD) δ 139.6, 134.3, 133.9, 130.2, 128.9, 127.4, 31.6, 21.8, 13.1.

trans-2-(5-Fluoro-2-methoxyphenyl)cyclopropylamine (6h). Method VI. A mixture of 5h (4.6 g, 21.9 mmol), triethylamine (4.27 mL, 30.6 mmol), and ethyl chloroformate (3.14 mL, 32.8 mmol) in dry acetone (150 mL) was stirred at -10 °C, and a solution of NaN₃ (2.42 g, 37.2 mmol) in H_2O (10 mL) was added after 2.5 h. The stirring was discontinued after an additional 2 h. The resulting suspension was poured into cold H_2O (220 mL) and was extracted with toluene (4 \times 100 The combined organic layers were dried (MgSO₄), mL). filtered, and concentrated to about 50% of the volume to remove remaining traces of H₂O. The resulting solution was heated (90 °C bath temperature) until the evolution of nitrogen ceased (about 3 h), and the solution was concentrated. The resulting isocyanate was dissolved in dry t-BuOH (50 mL), and the solution was refluxed for 23 h. The reaction mixture was concentrated, and the crude tert-butyl carbamate was purified on a silica column eluted with ethyl acetate/hexane (1:3). The carbamate was hydrolyzed by use of 1 M aqueous HCl (200 mL) at 100 °C (both temperature) for 15 h. The solution was washed with ether (3 \times 100 mL), alkalinized with saturated aqueous K_2CO_3 , and extracted with EtOAc (4 \times 200 mL). The combined organic layers were dried (K_2CO_3) , filtered, and concentrated. The oily primary amine was converted into the hydrochloride and recrystallized to afford pure 6h·HCl: ¹H NMR (CD₃OD) δ 6.95–6.93 (m, 2 H), 6.74 (d, J = 9.0 Hz, 1 H), 3.86 (s, 1 H), 2.79 (ddd, $J_1 = 3.6$ Hz, $J_2 = 4.2$ Hz, $J_3 = 7.8$ Hz, 1 H), 2.53 (ddd, $J_1 = 3.6$ Hz, $J_2 = 6.9$ Hz, $J_3 = 10.1$ Hz, 1 H), 1.38 (ddd, $J_1 = 4.2$ Hz, $J_2 = 6.7$ Hz, $J_3 = 10.1$ Hz, 1 H), 1.32 (ddd, $J_1 = 6.9$ Hz, $J_2 = 6.7$ Hz, $J_3 = 7.8$ Hz, 1 H); ¹³C NMR (CD₃OD) δ 158.3 (d, J = 237 Hz), 155.8 (d, J = 2.1 Hz), 129.6 (d, J = 7.7 Hz), 114.6 (d, J = 23.0 Hz).

(1*R*,2*R*)-*trans*-2-(2-Thienyl)cyclopropylamine [(1*R*,2*R*)-6j]. Method VII. Ethyl chloroformate (2.3 mL, 24 mmol) was added to a cooled solution (-10 °C) of (1*R*,2*R*)-5j (2.9 g, 17 mmol) and triethylamine (2.9 mL, 21 mmol) in dry acetone (100 mL). After 2 h, a solution of NaN₃ (1.68 g, 26 mmol) in H₂O (5 mL) was added to the stirred solution. The stirring was discontinued after 1 h, and H₂O (100 mL) was added. The solution was concentrated, and the remaining H₂O solution was extracted with ether (4×100 mL). The combined organic layers were dried (MgSO₄), filtered, and concentrated. The residue was dissolved in toluene (150 mL). The solution was concentrated to about half the volume in order to remove remaining traces of H₂O and was then heated (90 °C bath temperature) until the evolution of N₂ ceased (2 h). The volatiles were evaporated *in vacuo*. The resulting isocyanate was dissolved in dry 2-(trimethylsilyl)ethanol (5 mL), and the solution was heated to 60 °C (bath temperature) for 14 h. The crude carbamate was purified on a silica column eluted with ether/petroleum ether (1:2). The purified carbamate was treated with tetrabutylammonium fluoride (1 M solution in THF, 19 mL, 19 mmol) at 50 °C for 19 h. H₂O (50 mL) was added, and the stirring was discontinued after an additional 15 min. The mixture was concentrated, and the remaining water solution was acidified with 1 M aqueous HCl, washed with CH₂Cl₂ (50 mL), alkalinized [Na₂CO₃(s)], and extracted with EtOAc₃ (3×100 mL). The combined organic layers were dried (K₂CO₃) and concentrated. The crude primary amine was converted into the oxalate and recrystallized to afford (1R,2R)-**6j**·C₂O₄H₂: ¹H NMR (CD₃OD) δ 7.17 (dd, $J_1 = 1.3$ Hz, $J_2 = 5.1$ Hz, 1 H), 6.91–6.83 (m, 2 H), 2.74 (ddd, $J_1 = 3.5$ Hz, $J_2 = 4.4$ Hz, $J_3 = 7.9$ Hz, 1 H), 2.47 (dddd, $J_1 = 0.75$ Hz, $J_2 =$ $3.5 \text{ Hz}, J_3 = 6.4 \text{ Hz}, J_4 = 9.9 \text{ Hz}, 1 \text{ H}), 1.40 \text{ (ddd, } J_1 = 4.4 \text{ Hz},$ $J_2 = 6.6$ Hz, $J_3 = 9.9$ Hz, 1 H), 1.22 (ddd, $J_1 = 6.4$ Hz, $J_2 = 6.6$ Hz, $J_3 = 7.9$ Hz, 1 H); ¹³C NMR (CD₃OD) δ 166.7, 143.2, 128.0, 125.3, 124.8, 32.6, 17.8, 14.7.

trans-2-(5-Fluoro-2-methoxyphenyl)-N,N-dipropylcyclopropylamine (7h). Method VIII. A mixture of 6h·HCl (0.5 g, 2.3 mmol), 1-iodopropane (0.54 mL, 5.5 mmol), and finely ground K₂CO₃ (1.14 g, 8.3 mmol) in MeCN (10 mL) was stirred at room temperature for 4 days. Ether (15 mL) was added, and insoluble material was removed by filtration. The filtrate was concentrated, and the residue was chromatographed on an alumina column eluted with ether/petroleum ether (1:16). The resulting tertiary amine was converted into the hydrochloride and recrystallized to afford 7h·HCl as a white solid: ¹H NMR (CD₃OD, 90 MHz) δ 7.00–6.92 (m, 2 H), 6.75 (d, J = 9.4 Hz, 1 H), 3.88 (s, 1 H), 3.39–3.20 (m, 4 H), 3.00-2.64 (m, 2 H), 1.97-1.50 (m, 6 H), 1.03 (dd, $J_1 = 7.2$ Hz, $J_2 = 7.2$ Hz, 1 H); ¹³C NMR (CD₃OD) δ 158.4 (d, J = 237 Hz), 155.8 (d, J = 2.1 Hz), 128.7 (d, J = 7.7 Hz), 115.0 (d, J = 23.0Hz), 114.3 (d, J = 24.4 Hz), 112.6 (d, J = 8.4 Hz), 57.7 (2 C's), 56.5, 46.0, 18.5 (3 C's), 12.7, 11.5 (2C's).

trans-2-(5-Fluoro-2-hydroxyphenyl)-N,N-dipropylcyclopropylamine (71). Method IX. A stirred solution of 7h. HCl (0.15 g, 0.50 mmol) in 48% aqueous HBr (10 mL) was heated for 2 h (120 °C bath temperature). The solution was concentrated, and the residue was partitioned between saturated aqueous NaHCO₃ (20 mL) and ether (20 mL). The aqueous layer was extracted with ether (3 \times 30 mL), and the combined organic layers were dried (Na₂SO₄, filtered, and concentrated. The crude 71 was converted into the hydrochloride and recrystallized to afford 71·HCl as a white solid: ¹H NMR (CD₃OD, 90 MHz) δ 6.85–6.60 (m, 3 H), 3.42–3.23 (m, 4 H), 3.00-2.61 (m, 2 H), 1.98-1.50 (m, 6 H), 1.02 (dd, J₁ = 7.2 Hz, J_2 = 7.2 Hz, 6 H); ¹³C NMR (CD₃OD) δ 157.7 (d, J = 235 Hz), 153.6 (d, J = 1.4 Hz), 126.4 (d, J = 7.7 Hz), 116.4 (d, J = 8.3 Hz), 115.0 (d, J = 23.0 Hz), 114,2 (d, J = 24.3 Hz), 57.6 (2 C's), 45.8, 18.8, 18.4 (2 C's), 12.5, 11.3 (2 C's).

Pharmacology. Animals. Male Sprague-Dawley rats (B & K Universal, Sollentuna, Sweden), weighing 250–300 g, were used throughout the studies. The rats were housed five/ cage under standardized environmental conditions (temperature 22–23 °C; humidity 55–60%; 14/10 light/dark cycle, lights on at 6 am; rat chow and tap water allowed *ad lib*) for at least 1 week after the arrival until used in the experiments. Reserpine pretreatment (5 mg/kg, sc) was given to all animals 18–20 h before start of the experiments (cf. below).

Drugs. The compounds to be tested were dissolved in physiological (0.9%, w/v) saline immediately before use, occasionally with the addition of a few drops of glacial HOAc and/or moderate heating to obtain complete dissolution. Reserpine (Ciba, Basel, Switzerland) was dissolved in a few drops of glacial HOAc and made up to volume with 5.5% glucose (w/v). All drugs, including reserpine and NSD1015 (Sigma, St. Louis, MO), were subcutaneously injected into the neck region, in a volume of 5 mL/kg.

Biochemistry. The biochemical experiments, including brain dissections and HPLC determinations (electrochemical detection) of tissue contents of DOPA and 5-HTP, were carried out essentially according to methods detailed elsewhere.^{25,26} Empirically, the maximum decrease of cerebral 5-HTP values

obtained with a direct-acting 5-HT_{1A} receptor agonist (*e.g.*, **3**) is about 50% from control values under conditions equivalent to those used in the present experiments (see ref 4c).

Behavior. The occurrence and intensity of flattened body posture, forepaw treading, and hindlimb abduction was scored for 60 s at the 30-min time point following test drug administration, using an intensity-based rating scale, where 0 = absent, 1 = equivocal, 2 = definite, and 3 = intense;²⁷ *i.e*, the maximum summed score for the total set of behavioral items was 9. For reference, a dose of $0.1 \,\mu$ mol/kg (sc) of (*R*)-**3** which causes a near-maximal postsynaptic 5-HT_{1A} behavioral activation yields summed scores 8-9 using this approach under similar conditions (cf. ref 27). Observations of other effects of the drugs on behavior were also simultaneously noted, though not quantified.

Statistics. The biochemical data are expressed in percent of the corresponding control values, means \pm SEM. One-way ANOVA, followed by Fisher's protected least-significant-difference test (PLSD), was used for statistical comparisons of drug-treated groups with controls. The behavioral data are expressed as medians (range). Probability levels \leq 5% were considered statistically significant.

5-HT_{1A} Receptor Binding Assay. Male Sprague–Dawley rats (weighing about 200 g) were decapitated, and the cortex and hippocampus were dissected. The tissues (600-900 mg) from each rat were immediately homogenized in 15 mL of icecold 50 mM Tris-HCl buffer containing 4.0 mM CaCl₂ and 5.7 mM ascorbic acid, pH 7.5, with an Ultra Turrax (Janke and Kunkel, Staufen, FRG) for 10 s. After centrifugation for 12.5 min at 17000 rpm (39800g) in a Beckman centrifuge with a chilled JA-17 rotor (Beckman, Paulo Alto, CA), the pellets were resuspended in the same buffer, and homogenization and centrifugation were repeated. The pellets from at least six rats were again suspended in the buffer, pooled, homogenized, and stored on ice for 1–4 h. The tissue homogenate was diluted to 10 mg/1.25 mL with the buffer, incubated for 10 min at 37 °C, and supplied with 10 μ M pargylin (Sigma, St. Louis, MO) followed by reincubation for 10 min.

Incubation mixtures (2 mL) contained 1-300 nM test compound (diluted in 50 mM Tris-HCl containing 5.7 mM ascorbic acid, pH 7.5), 2 nM [3H]-8-OH-DPAT ([3H]-8-hydroxy-2-(dipropylamino)tetralin hydrobromide), 31.60 Ci/mmol, New England Nuclear, Dreieich, Germany, and Research Biochemicals, Wayland, MA), 5 mg/mL tissue homogenate in 50 mM Tris-HCl buffer containing 4.0 mM CaCl₂ and 5.7 mM ascorbic acid, pH 7.5. Binding experiments were started by the addition of tissue homogenate followed by incubation at 37 °C for 10 min. The incubation mixtures were filtered through Whatman GF/B glass filters with a Brandel Cell Harvester (Gaithersburg, MD). The filters were washed twice with 5 mL of ice-cold 50 mM Tris-HCl buffer, pH 7.5, and counted with 5 mL of Packard Ultima Gold in a Beckman LS 3801 scintillation counter. Nonspecific binding was measured by the addition of 10 μ M 5-HT·HCl to the reaction mixture. The binding data were processed by nonlinear least squares computer analysis.²⁸ A K_d value of 1.4 nM for 8-OH-DPAT binding was obtained from the saturation experiments and was used to calculate the K_i values. The range of K_i was calculated from the inverse of the standard error of the estimate of K_{i} . Four of the compounds in Table 4 [(1*S*,2*R*)-7h, (1*R*,2*S*)-7h, (1*S*,2*R*)-7l, and (1R, 2S)-71] were assayed for 5-HT_{1A} binding using a slightly modified procedure.²⁹

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